

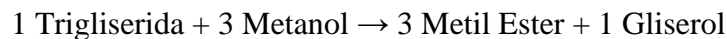
## LAMPIRAN

### A. PERHITUNGAN

#### 1. Perhitungan Jumlah Metanol

Contoh perhitungan jumlah metanol yang digunakan dalam setiap tahap dihitung sebagai berikut:

Perbandingan molar minyak jelantah dengan metanol	= 1:6
Volume minyak jelantah	= 100 mL
Densitas metanol ( $\rho$ metanol)	= 0,788 g/mL
Densitas minyak jelantah ( $\rho$ minyak jelantah)	= 0,9102 g/mL
Berat molekul minyak jelantah (BM minyak jelantah)	= 850,32 g/mol
Berat molekul metanol (BM metanol)	= 32 g/mol



- Menentukan mol minyak jelantah

$$\text{Massa minyak} = V \times \rho \text{ minyak jelantah}$$

$$\text{Massa minyak} = 100 \text{ mL} \times 0,9102 \text{ g/mL}$$

$$\text{Massa minyak} = 91,02 \text{ gram}$$

$$\text{Mol minyak} = (\text{massa minyak})/(\text{Berat Molekul minyak})$$

$$\text{Mol minyak} = (91,02 \text{ g})/(850,32 \text{ g/mol})$$

$$\text{Mol minyak} = 0,107 \text{ mol}$$

- Menentukan mol metanol

$$\text{Mol metanol} = (\text{koef metanol})/(\text{koef minyak}) \times \text{mol minyak}$$

$$\text{Mol metanol} = 3/1 \times 0,107 \text{ mol}$$

Mol metanol = 0,3211 mol

Perbandingan 6:1 = 0,3211 mol  $\times$  6 = 1,9266 mol

- Menentukan volume metanol

Massa metanol = mol metanol  $\times$  BM metanol

Massa metanol = 1,9266 mol  $\times$  32 g/mol

Massa metanol = 61,6512 gram

Volume metanol = (massa metanol)/( $\rho$  metanol)

Volume metanol = ( 61,6512 gram )/(0,788 g/mL)

Volume metanol = 78,2376 mL = 78 mL

## 2. Perhitungan Jumlah Katalis

Katalis abu kulit pisang kepok matang yang digunakan dalam reaksi transesterifikasi sebanyak 1% berat minyak. Contoh perhitungan kebutuhan katalis adalah sebagai berikut:

% katalis = 1% berat minyak jelantah

$\rho$  minyak jelantah = 0,9102 g/mL

Massa katalis =  $\rho$ minyak jelantah  $\times$   $V$ minyak yang digunakan  $\times$  % katalis (b/b)

Massa katalis = 0,9102 gr/mL  $\times$  100 mL  $\times$  1%

Massa katalis = 0,9102 gram (untuk 1% berat minyak)

Massa katalis = 0,9102 gram (berat katalis untuk 1 variasi)

Total untuk 27 variasi = 27  $\times$  0,9102 gram = 24,575 gram

## 3. Perhitungan Bilangan Asam Lemak Bebas (FFA)

Perhitungan asam lemak bebas minyak jelantah adalah sebagai berikut:

- ✓ Minyak Jelantah sebelum *Pretreatment*

$$\%FFA = (N_{\text{NaOH}} \times V_{\text{NaOH}} \times \text{BM}_{\text{asam lemak}}) / (\text{massa sampel} \times 1000) \times 100\%$$

$$\%FFA = (0,25 \text{ N} \times 1,8 \text{ ml} \times 282 \text{ g/mol}) / (7,05 \text{ gram} \times 1000) \times 100\%$$

$$\%FFA = 1,8 \%$$

✓ Minyak Jelantah setelah *Pretreatment*

$$\%FFA = (N_{\text{NaOH}} \times V_{\text{NaOH}} \times \text{BM}_{\text{asam lemak}}) / (\text{massa sampel} \times 1000) \times 100\%$$

$$\%FFA = (0,25 \text{ N} \times 1,25 \text{ ml} \times 282 \text{ g/mol}) / (7,05 \text{ gram} \times 1000) \times 100\%$$

$$\%FFA = 1,25 \%$$

✓ Penurunan Presentase ALB (*Asam Lemak Bebas*)

$$\% \text{ Penurunan ALB (Asam Lemak Bebas) } = ( \%FFA_{\text{sebelum pretreatment}} - \%FFA_{\text{setelah pretreatment}} ) / \%FFA_{\text{sebelum pretreatment}} \times 100$$

$$= ((1,8 - 1,25) / 1,8) \times 100$$

$$= 30,56 \%$$

#### 4. Perhitungan Densitas

Contoh perhitungan densitas minyak jelantah dan biodiesel sebagai berikut:

Berat Piknometer kosong ( $m_{\text{piknokosong}}$ ) = 26,7 gram

Volume piknometer ( $V_{\text{piknometer}}$ ) = 50 ml

✓ Densitas Minyak Jelantah sebelum *Pretreatment*

Pada kondisi suhu ruang (26 °C)

Berat piknometer kosong + minyak jelantah = 73,12 gram

Berat Minyak = Berat<sub>piknometer kosong + minyak jelantah</sub> – Berat<sub>piknometer kosong</sub>

$$= 46,42 \text{ gram}$$

Densitas minyak jelantah = (massa minyak jelantah) / (Volume piknometer)

Densitas minyak jelantah = (46,42 gram) / 50 ml

Densitas minyak jelantah (26 °C) = 0,9284 gr/ml = 928,4 kg/m<sup>3</sup>

Pada kondisi direndam air (40 °C)

Berat piknometer kosong + minyak jelantah = 68,42 gram

Berat Minyak = Berat<sub>piknometer kosong + minyak jelantah</sub> - Berat<sub>piknometer kosong</sub>  
= 41,72 gram

Densitas minyak jelantah = (massa minyak jelantah)/(Volume piknometer)

Densitas minyak jelantah = (41,72 gram)/50 ml

Densitas minyak jelantah (40 °C) = 0,8344 gr/ml = 834,4 kg/m<sup>3</sup>

✓ Densitas Minyak Jelantah setelah *Pretreatment*

Pada kondisi suhu ruang (26 °C)

Berat piknometer kosong + minyak jelantah = 72,21 gram

Berat Minyak = Berat<sub>piknometer kosong + minyak jelantah</sub> - Berat<sub>piknometer kosong</sub>  
= 45,51 gram

Densitas minyak jelantah = (massa minyak jelantah)/(Volume piknometer)

Densitas minyak jelantah = (45,51 gram)/50 ml

Densitas minyak jelantah (26 °C) = 0,9102 gr/ml = 910,2 kg/m<sup>3</sup>

Pada kondisi suhu ruang (40 °C)

Berat piknometer kosong + minyak jelantah = 65,98 gram

Berat Minyak = Berat<sub>piknometer kosong + minyak jelantah</sub> - Berat<sub>piknometer kosong</sub>  
= 39,28 gram

Densitas minyak jelantah = (massa minyak jelantah)/(Volume piknometer)

Densitas minyak jelantah = (39,28 gram)/50 ml

Densitas minyak jelantah (40 °C) = 0,7856 gr/ml = 785,6 kg/m<sup>3</sup>

✓ Densitas Biodiesel

Berat piknometer kosong + biodiesel = 69,7 gram

$$\begin{aligned} \text{Berat Minyak} &= \text{Berat}_{\text{piknometer kosong + minyak jelantah}} - \text{Berat}_{\text{piknometer kosong}} \\ &= 43 \text{ gram} \end{aligned}$$

Densitas biodiesel = (massa biodiesel)/(Volume piknometer)

$$\text{Densitas biodiesel} = (43 \text{ gram}) / 50 \text{ ml}$$

$$\text{Densitas biodiesel} = 0,86 \text{ gr/ml} = 860 \text{ kg/m}^3$$

6. Perhitungan Viskositas Kinematik

Menggunakan *viscometer otswaldz*

Contoh perhitungan viskositas minyak jelantah dan biodiesel sebagai berikut:

$$\text{Viskositas sampel} = k \times \text{sg} \times t$$

Keterangan : k = konstanta viskosimeter (kg/m.s<sup>2</sup>)

$$t = \text{waktu alir (s)}$$

$$\text{Sg} = (\text{densitas sampel})/(\text{densitas air})$$

▪ Kalibrasi Air :

$$\rho_{\text{air}} (40 \text{ }^\circ\text{C}) = 992,2 \text{ kg/m}^3$$

$$\text{Viskositas air (40 }^\circ\text{C)} = 0,656 \times 10^{-3} \text{ kg/m.s}$$

$$\begin{aligned} \text{Sg}_{\text{air}} &= (\text{densitas air } 40 \text{ }^\circ\text{C})/(\text{densitas air } 4 \text{ }^\circ\text{C}) \\ &= 1 \text{ gr/ml} / 1 \text{ g/cm}^3 \end{aligned}$$

$$\text{Sg}_{\text{air}} = 1$$

$$t_{\text{air}} = 180 \text{ sekon}$$

$$\text{Viskositas air} = k \times \text{sg} \times t_{\text{air}}$$

$$0,656 \times 10^{-3} \text{ kg/m.s} = k \times 1 \times 180 \text{ sekon}$$

Suhu °C	Rapat massa ρ (kg/m <sup>3</sup> )	Viskositas Dinamik μ (Nd/m <sup>2</sup> )
0,0	999,9	1,792×10 <sup>-3</sup>
5,0	1000	1,519
10,0	999,7	1,308
20,0	998,2	1,005
30,0	995,7	0,801
40,0	992,2	0,656
50,0	988,1	0,549
60,0	983,2	0,469
70,0	977,8	0,406
80,0	971,8	0,357
90,0	965,3	0,317
100,0	958,4	0,284×10 <sup>-3</sup>

$$k = 0,656 \times 10^{-3} \text{ kg/m.s} / 180 \text{ s}$$

$$k = 3,64 \times 10^{-6} \text{ kg/m.s}^2$$

▪ Viskositas Minyak Jelantah

✓ sebelum *Pretreatment*

$$t_{\text{minyak jelantah}} = 12010 \text{ sekon}$$

$$\rho_{\text{minyak jelantah}} (40 \text{ }^\circ\text{C}) = 834,4 \text{ kg/m}^3$$

$$Sg_{\text{minyak jelantah}} = (\text{densitas sampel } 40 \text{ }^\circ\text{C}) / (\text{densitas air } 40 \text{ }^\circ\text{C})$$

$$= (834,4 \text{ kg/m}^3) / (992,2 \text{ kg/m}^3)$$

$$= 0,84$$

$$\text{Viskositas dinamik} = k \times sg \times t_{\text{biodiesel}}$$

$$\text{Minyak Jelantah} = 3,64 \times 10^{-6} \text{ kg/m.s}^2 \times 0,84 \times 12010 \text{ s}$$

$$\text{sebelum } \textit{Pretreatment} = 0,0368 \text{ kg/m.s}$$

$$\text{Viskositas kinematik minyak jelantah} = 0,0368 \text{ kg/m.s} / 834,4 \text{ kg/m}^3$$

$$(40 \text{ }^\circ\text{C}) = 39,65 \times 10^{-6} \text{ m}^2/\text{s} = 39,65 \text{ mm}^2/\text{s} \text{ (cSt)}$$

✓ setelah *Pretreatment*

$$t_{\text{minyak jelantah}} = 12000 \text{ sekon}$$

$$\rho_{\text{minyak jelantah}} (40 \text{ }^\circ\text{C}) = 785,6 \text{ kg/m}^3$$

$$Sg_{\text{minyak jelantah}} = (\text{densitas sampel } 40 \text{ }^\circ\text{C}) / (\text{densitas air } 40 \text{ }^\circ\text{C})$$

$$= (785,6 \text{ kg/m}^3) / (992,2 \text{ kg/m}^3)$$

$$= 0,79$$

$$\text{Viskositas dinamik} = k \times sg \times t_{\text{biodiesel}}$$

$$\text{Minyak Jelantah} = 3,64 \times 10^{-6} \text{ kg/m.s}^2 \times 0,79 \times 12000 \text{ s}$$

$$\text{setelah } \textit{Pretreatment} = 0,0346 \text{ kg/m.s}$$

Viskositas kinematik minyak jelantah =  $0,0346 \text{ kg/m.s} / 785,6 \text{ kg/m}^3$

$$(40 \text{ }^\circ\text{C}) = 38,04 \times 10^{-6} \text{ m}^2/\text{s} = 38,04 \text{ mm}^2/\text{s} \text{ (cSt)}$$

▪ Viskositas Biodiesel

$t_{\text{biodiesel}} = 700 \text{ sekon}$

$\rho_{\text{biodiesel}} (40 \text{ }^\circ\text{C}) = 860 \text{ kg/m}^3$

$Sg_{\text{biodiesel}} = (\text{densitas sampel } 40 \text{ }^\circ\text{C})/(\text{densitas air } 40 \text{ }^\circ\text{C})$

$$= (860 \text{ kg/m}^3)/(992,2 \text{ kg/m}^3)$$

$$= 0,87$$

Viskositas dinamik biodiesel =  $k \times sg \times t_{\text{biodiesel}}$

$$= 3,64 \times 10^{-6} \text{ kg/m.s}^2 \times 0,87 \times 700 \text{ s}$$

$$= 0,0022 \text{ kg/m.s}$$

Viskositas kinematik biodiesel =  $0,0022 \text{ kg/m.s} / 860 \text{ kg/m}^3$

$$(40 \text{ }^\circ\text{C}) = 2,57 \times 10^{-6} \text{ m}^2/\text{s} = 2,57 \text{ mm}^2/\text{s} \text{ (cSt)}$$

## 6. Perhitungan Yield Biodiesel

Contoh perhitungan yield biodiesel sebagai berikut:

Massa Minyak Jelantah = 91,02 gram

Yield Biodiesel =  $(\text{massa biodiesel})/(\text{massa minyak}) \times 100 \%$

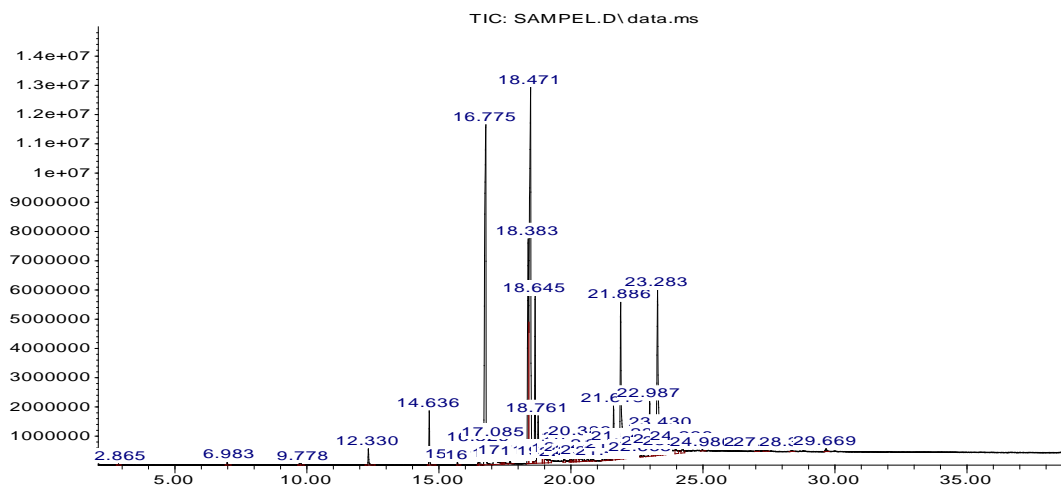
Yield Biodiesel =  $82,77 \text{ gram} / 91,02 \text{ gram} \times 100 \%$

Yield Biodiesel = 90,93 %

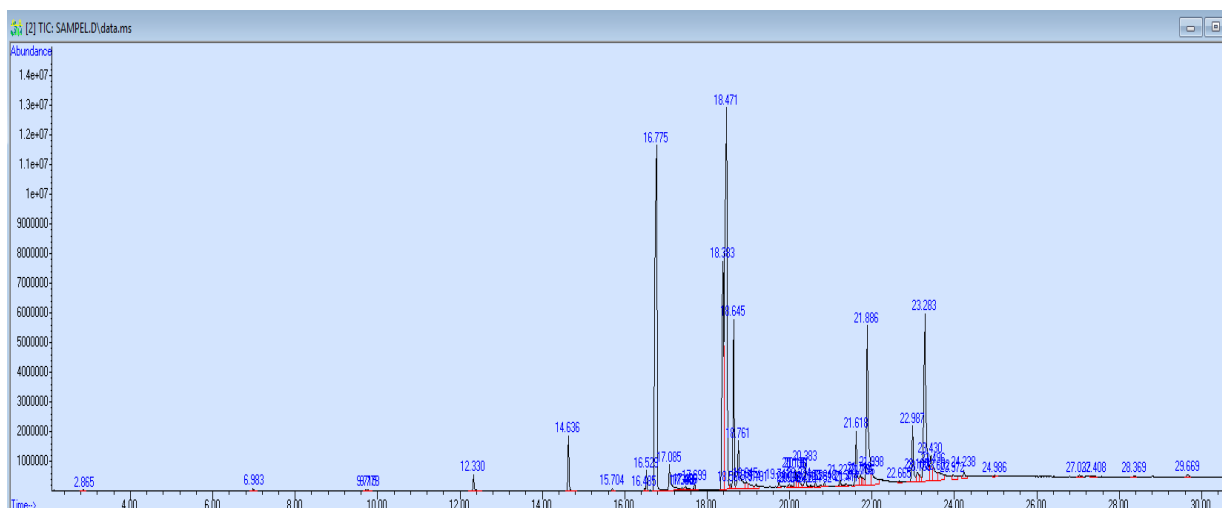
## B. HASIL ANALISA

- GC-MS

Abundance



Time-->



### Library Search Report

Data Path : D:\MAHASISWA\UMI HANIFAH UNTIRTA\  
Data File : SAMPEL.D  
Acq On : 01 Nov 2022 15:06  
Operator : UMI HANIFAH  
Sample : SAMPEL BIODIESEL  
Misc : S1 UNIV SULTAN AGENG TIRTAYASA  
ALS Vial : 2 Sample Multiplier: 1

Search Libraries: D:\MassHunter\...ry\WILLEY09TH.L Minimum Quality: 0  
Unknown Spectrum: Apex  
Integration Events: ChemStation Integrator - events.e

Pk#	RT	Area%	Library/ID	Ref#	CAS#	Qual
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1	2.865	0.03	D:\MassHunter\Library\WILLEY09TH.L Hexanal (CAS) \$\$ n-Hexanal \$\$ Hexa	10655	000066-25-1	64
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 aldehyde  
 Hexanal (CAS) \$\$ n-Hexanal \$\$ Hexa 10650 000066-25-1 64  
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 aldehyde  
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 xanal \$\$ Caproaldehyde \$\$ Caproic  
 aldehyde

2 6.983 0.07 D:\MassHunter\Library\WILLEY09TH.L  
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 caprylate  
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 \$\$ Methyl octanoate \$\$ Methyl capr  
 ylate  
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 2,4-Decadienal, (E,E)- (CAS) \$\$ tr 70578 025152-84-5 95  
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 cadien-1-al  
 2,4-Decadienal, (E,E)- (CAS) \$\$ tr 70575 025152-84-5 94  
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 cadien-1-al

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 Decanoic acid, methyl ester \$\$ Cap 145189 000110-42-9 95  
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 2095  
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 canoate  
 Dodecanoic acid, methyl ester (CAS 218640 000111-82-0 97  
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 canoate  
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 Tetradecanoic acid, methyl ester ( 298306 000124-10-7 97

CAS) \$\$ Methyl myristate \$\$ Uniph  
t A50  
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CAS) \$\$ Methyl myristate \$\$ Uniph  
t A50

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Methyl 13-methyltetradecanoate 337844 999337-84-4 99  
Pentadecanoic acid, methyl ester \$ 337746 007132-64-1 99  
\$ Methyl n-pentadecanoate  
Pentadecanoic acid, methyl ester ( 337751 007132-64-1 98  
CAS) \$\$ Methyl pentadecanoate

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9-Hexadecenoic acid, methyl ester, 371355 001120-25-8 99  
(Z)- (CAS) \$\$ Methyl palmitoleate  
9-Hexadecenoic acid, methyl ester, 371346 001120-25-8 99  
(Z)- \$\$ Methyl palmitoleate  
9-Hexadecenoic acid, methyl ester, 371351 001120-25-8 99  
(Z)- (CAS) \$\$ Methyl palmitoleate

9 16.529 1.49 D:\MassHunter\Library\WILLEY09TH.L  
Methyl hexadec-9-enoate 371471 010030-74-7 99  
9-Hexadecenoic acid, methyl ester, 371342 001120-25-8 99  
(Z)- (CAS) \$\$ Methyl palmitoleate  
9-Hexadecenoic acid, methyl ester, 371351 001120-25-8 99  
(Z)- (CAS) \$\$ Methyl palmitoleate

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AS) \$\$ Methyl palmitate \$\$ Uniphat  
A60  
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AS) \$\$ Methyl palmitate \$\$ Uniphat  
A60  
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Palmitic acid, methyl ester \$\$ Un  
iphat A60

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n-Hexadecanoic acid \$\$ Hexadecanoi 337724 000057-10-3 99  
c acid \$\$ n-Hexadecoic acid \$\$ Pal  
mitic acid  
Hexadecanoic acid (CAS) \$\$ Palmiti 337729 000057-10-3 99  
c acid \$\$ Palmitinic acid \$\$ Prifr  
ac 2960  
Hexadecanoic acid (CAS) \$\$ Palmiti 337728 000057-10-3 99  
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ac 2960

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ac 2960  
Octadecanoic acid (CAS) \$\$ Stearic 415832 000057-11-4 86  
acid \$\$ n-Octadecanoic acid \$\$ Va  
nicol  
Octadecanoic acid (CAS) \$\$ Stearic 415828 000057-11-4 60

acid \$\$ n-Octadecanoic acid \$\$ Vanicolic acid

13 17.445 0.08 D:\MassHunter\Library\WILLEY09TH.L  
Hexadecanoic acid (CAS) \$\$ Palmitic acid 337720 000057-10-3 89  
c acid \$\$ Palmitic acid \$\$ Prifrac 2960  
Hexadecanoic acid (CAS) \$\$ Palmitic acid 337729 000057-10-3 83  
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cis-Vaccenic acid 410259 000506-17-2 50

14 17.488 0.17 D:\MassHunter\Library\WILLEY09TH.L  
cis-10-Heptadecenoic acid, methyl ester 410261 999410-26-1 99  
CIS-10-HEPTADECENOIC ACID ME 410223 999410-22-3 91  
9-Octadecenoic acid (Z)-, methyl ester (CAS) \$\$ Methyl oleate 447935 000112-62-9 90

15 17.699 0.49 D:\MassHunter\Library\WILLEY09TH.L  
Heptadecanoic acid, methyl ester (CAS) \$\$ Methyl heptadecanoate 415858 001731-92-6 98  
Heptadecanoic acid, methyl ester (CAS) \$\$ Methyl heptadecanoate 415850 001731-92-6 97  
Hexadecanoic acid, 15-methyl-, methyl ester \$\$ Methyl isoheptadecanoate 415885 006929-04-0 97

16 18.383 9.60 D:\MassHunter\Library\WILLEY09TH.L  
9,12-Octadecadienoic acid (Z,Z)-, methyl ester (CAS) \$\$ Methyl linoleate 442621 000112-63-0 99  
9,12-Octadecadienoic acid, methyl ester 442673 002462-85-3 99  
9,12-Octadecadienoic acid (Z,Z)-, methyl ester (CAS) \$\$ Methyl linoleate 442618 000112-63-0 99

17 18.471 33.33 D:\MassHunter\Library\WILLEY09TH.L  
9-Octadecenoic acid, methyl ester, (E)- \$\$ Elaidic acid, methyl ester 447972 001937-62-8 99  
9-Octadecenoic acid (Z)-, methyl ester (CAS) \$\$ Methyl oleate 447957 000112-62-9 99  
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18 18.561 0.38 D:\MassHunter\Library\WILLEY09TH.L  
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19 18.645 6.88 D:\MassHunter\Library\WILLEY09TH.L  
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ester 9718  
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 9718  
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 9718

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 9-OCTADECENOIC ACID \$\$ ELAIDINSAEU 410099 999410-09-9 99  
 RE  
 OCTADEC-9-ENOIC ACID \$\$ 9-OCTADECE 410104 999410-10-4 99  
 NOIC ACID  
 9-Octadecenoic acid, (E)- \$\$ trans 410193 000112-79-8 99  
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21 18.945 0.04 D:\MassHunter\Library\WILLEY09TH.L  
 Thiosulfuric acid (H2S2O3), S-(2-a 80762 002937-53-3 96  
 minoethyl) ester \$\$ Cysteamine, S-  
 sulfo-  
 9-Octadecenoic acid (Z)- (CAS) \$\$ 410112 000112-80-1 95  
 Oleic acid \$\$ Red oil \$\$ Oelsauere  
 OCTADEC-9-ENOIC ACID \$\$ 9-OCTADECE 410107 999410-10-7 95  
 NOIC ACID

22 19.014 0.56 D:\MassHunter\Library\WILLEY09TH.L  
 9-Octadecenoic acid (Z)- (CAS) \$\$ 410108 000112-80-1 99  
 Oleic acid \$\$ Red oil \$\$ Oelsauere  
 9-Octadecenoic acid (Z)- (CAS) \$\$ 410102 000112-80-1 99  
 Oleic acid \$\$ Red oil \$\$ Oelsauere  
 6-Octadecenoic acid 410222 999410-22-2 99

23 19.191 0.63 D:\MassHunter\Library\WILLEY09TH.L  
 Methyl 9-cis,11-trans-octadecadien 442583 999442-58-3 95  
 oate  
 9,17-Octadecadienal, (Z)- \$\$ (9Z)- 360379 056554-35-9 94  
 9,17-Octadecadienal #  
 9,12-Octadecadienoic acid (Z,Z)-, 442635 000112-63-0 93  
 methyl ester (CAS) \$\$ Methyl linol  
 eate

24 19.741 0.05 D:\MassHunter\Library\WILLEY09TH.L  
 2-(METHYL-D3)-CYCLONONANONE \$\$ Cyc 76587 032454-54-9 55  
 lononanone, 2-methyl-d3- (CAS)  
 1-Heptadecene (CAS) \$\$ Hexahydroap 287872 006765-39-5 43  
 lotaxene \$\$ 1 - heptadecene  
 cis-Vaccenic acid 410259 000506-17-2 41

25 19.963 0.17 D:\MassHunter\Library\WILLEY09TH.L  
 10,13-Octadecadienoic acid, methyl 442595 056554-62-2 90  
 ester (CAS)  
 Dibutylphosphoramidate 203924 999203-92-4 55  
 Cyclopropaneoctanal, 2-octyl- \$\$ 8 404888 056196-06-6 53  
 -(2-Octylcyclopropyl)octanal #

26 20.003 0.03 D:\MassHunter\Library\WILLEY09TH.L  
 Cyclopentadecanone, 2-hydroxy- \$\$ 292764 004727-18-8 89  
 2-Hydroxycyclopentadecanone #

9-Octadecenoic acid (Z)- (CAS) \$\$ 410112 000112-80-1 60  
 Oleic acid \$\$ Red oil \$\$ Oelsauere  
 cis-10-Heptadecenoic acid, methyl 410261 999410-26-1 59  
 ester

27 20.046 0.04 D:\MassHunter\Library\WILLEY09TH.L  
 Fumaric acid, tetradec-3-enyl unde 726090 999726-09-0 38  
 cyl ester  
 cis-11-Eicosenoic acid, methyl est 518835 999518-83-5 38  
 er  
 7-Isopropyl-4a-methyloctahydro-2(1 203116 054594-42-2 30  
 H)-naphthalenone #

28 20.136 0.05 D:\MassHunter\Library\WILLEY09TH.L  
 2-(METHYL-D3)-CYCLONONANONE \$\$ Cyc 76587 032454-54-9 38  
 lononanone, 2-methyl-d3- (CAS)  
 Palmitoyl chloride \$\$ Hexadecanoyl 387805 000112-67-4 38  
 -chloride- \$\$ Palmitic acid chlori  
 de  
 Palmitoyl chloride \$\$ Hexadecanoyl 387803 000112-67-4 38  
 -chloride- \$\$ Palmitic acid chlori  
 de

29 20.181 0.79 D:\MassHunter\Library\WILLEY09TH.L  
 cis-11-Eicosenoic acid, methyl est 518835 999518-83-5 99  
 er  
 methyl ester of cis-.DELTA.(11)-C( 518838 003946-08-5 94  
 20:1) acid \$\$ CIS-11-EICOSENOIC AC  
 IDME  
 cis-13-Eicosenoic acid, methyl est 518901 999518-90-1 93  
 er

30 20.234 0.06 D:\MassHunter\Library\WILLEY09TH.L  
 4-(Trifluoromethyl)phenylacetone 142256 002338-75-2 35  
 nitrile \$\$ p-Trifluoromethylphenylacet  
 onitrile  
 Adipic acid, nonyl trans-2-methylc 609715 999609-71-5 27  
 yclohexyl ester  
 cis 2-methyl 1-phenyl 1-cyclopenta 143016 039522-58-2 27  
 necarbonitrile

31 20.383 1.86 D:\MassHunter\Library\WILLEY09TH.L  
 Eicosanoic acid, methyl ester (CAS 523614 001120-28-1 99  
 ) \$\$ Arachidic acid methyl ester  
 Eicosanoic acid, methyl ester \$\$ M 523616 001120-28-1 99  
 ethyl arachisate \$\$ Methyl eicosan  
 oate  
 Eicosanoic acid, methyl ester (CAS 523618 001120-28-1 99  
 ) \$\$ Arachidic acid methyl ester

32 20.511 0.01 D:\MassHunter\Library\WILLEY09TH.L  
 1H-Indole, 2-methyl-3-phenyl- (CAS 199687 004757-69-1 43  
 ) \$\$ 3 PHENYL-2-METHYLINDOLE  
 2-Myristinoyl-glycinamide 403683 999403-68-3 27  
 1,1,1,3,5,5,5-Heptamethyltrisiloxa 238341 001873-88-7 25  
 ne \$\$ Bis(trimethylsiloxy)methylsi  
 lane

33 20.633 0.19 D:\MassHunter\Library\WILLEY09TH.L  
Octadec-9-enoic acid 410106 999410-10-6 42  
OCTADEC-9-ENOIC ACID \$\$ 9-OCTADECE 410107 999410-10-7 42  
NOIC ACID  
9-Methyl-Z-10-tetradecen-1-ol acet 371406 999371-40-6 38  
ate

34 20.752 0.03 D:\MassHunter\Library\WILLEY09TH.L  
Pyridine-3-carboxamide, oxime, N-( 405484 288246-53-7 59  
2-trifluoromethylphenyl)-  
3-Quinolinecarboxylic acid, 6,7-di 327604 999327-60-4 55  
fluoro-1,4-dihydro-4-oxo-, ethyles  
ter  
1,2-Benzisothiazol-3-amine tbdms 358158 999358-15-8 38

35 20.848 0.01 D:\MassHunter\Library\WILLEY09TH.L  
1H-Indole, 2-methyl-3-phenyl- (CAS 199687 004757-69-1 49  
) \$\$ 3 PHENYL-2-METHYLINDOLE  
cis-11-Eicosenoic acid 484531 005561-99-9 47  
1,2,5-Oxadiazol-3-amine, 4-(3-meth 198339 999198-33-9 30  
oxyphenoxy)-

36 21.227 0.33 D:\MassHunter\Library\WILLEY09TH.L  
Bicyclo[4.3.1]decan-10-one 71209 020440-21-5 83  
Oleic acid, 3-hydroxypropyl ester 555014 000821-17-0 70  
\$\$ 3-Hydroxypropyl (9E)-9-octadece  
noate #  
9-Octadecenoic acid (Z)-, 2-hydrox 587486 003443-84-3 70  
y-1-(hydroxymethyl)ethyl ester \$\$  
2-Monoolein

37 21.374 0.01 D:\MassHunter\Library\WILLEY09TH.L  
Methyl 2-octylcyclopropene-1-octan 479341 003220-60-8 44  
oate  
Methyl 9,10-methylene-octadec-9-en 479340 999479-34-0 43  
oate  
Benzo[h]quinoline, 2,4-dimethyl- \$ 199727 000605-67-4 38  
\$ 2,4-Dimethylbenzo[h]quinoline #

38 21.417 0.01 D:\MassHunter\Library\WILLEY09TH.L  
2-Methyl-5H-dibenz[b,f]azepine 199741 999199-74-1 43  
1,2-Bis(trimethylsilyl)benzene \$\$ 240171 017151-09-6 42  
Trimethyl[2-(trimethylsilyl)phenyl  
]silane #  
1,1,1,3,5,5,5-Heptamethyltrisiloxa 238341 001873-88-7 38  
ne \$\$ Bis(trimethylsiloxy)methylsi  
lane

39 21.618 0.07 D:\MassHunter\Library\WILLEY09TH.L  
Ether, isopropyl 2-benzyl-2-propen 154812 999154-81-2 47  
yl  
1-Naphthalenol, 1,2,3,4-tetrahydro 154279 021503-12-8 43  
-, acetate \$\$ 1-Acetoxytetralin  
4-Phosphorinanone, 1-methyl- (CAS) 36262 016327-48-3 38  
\$\$ 1-Methyl-4-phosphorinanone

40 21.718 0.48 D:\MassHunter\Library\WILLEY09TH.L  
2-Butoxy-4-methyl-[1,3,2]dioxabori 112059 003208-68-2 30

nane  
Adipic acid, 2-ethylhexyl isobutyl ester 494032 999494-03-2 25  
2H-Azepine-2-thione, hexahydro- (C AS) \$\$ Thiocaprolactam 35231 007203-96-5 25

41 21.756 0.06 D:\MassHunter\Library\WILLEY09TH.L  
2-Naphthalene-sulfonic acid 200544 999200-54-4 53  
2,4-DIPHENYL GLUTARONITRILE 309450 999309-45-0 27  
1,3-Dihydrobenzo(c)thiophene-1-pen tanol 2,2dioxide 330194 072938-79-5 27

42 21.796 0.01 D:\MassHunter\Library\WILLEY09TH.L  
2-Methyl-5H-dibenz[b,f]azepine 199741 999199-74-1 53  
1,3-Dihydrobenzo(c)thiophene-1-pen tanol 2,2dioxide 330194 072938-79-5 22  
2-isopropylthio-N-allylaniline \$\$ N-Allyl-2-isopropylthioaniline 199393 102968-92-3 22

43 21.886 0.01 D:\MassHunter\Library\WILLEY09TH.L  
2-Amino-3-(.alpha.-furyl)-1,4-naphthoquinone 289084 999289-08-4 59  
3-Pyridinemethanamine, N-[[4-(dimethylamino)phenyl]methylidene]- 289462 999289-46-2 38  
Palmitoyl chloride \$\$ Hexadecanoyl chloride- \$\$ Palmitic acid chloride 387803 000112-67-4 35

44 21.998 1.36 D:\MassHunter\Library\WILLEY09TH.L  
Docosanoic acid, methyl ester (CAS ) \$\$ Methyl behenate \$\$ Methyl docosanoate 584204 000929-77-1 91  
Docosanoic acid, methyl ester \$\$ Behenic acid, methyl ester \$\$ Methyl behenate 584197 000929-77-1 78  
Docosanoic acid, methyl ester (CAS ) \$\$ Methyl behenate \$\$ Methyl docosanoate 584202 000929-77-1 68

45 22.665 0.1 D:\MassHunter\Library\WILLEY09TH.L  
2-Methyl-5H-dibenz[b,f]azepine 199741 999199-74-1 50  
Cyclotrisiloxane, hexamethyl- \$\$ Dimethylsiloxane cyclic trimer 238255 000541-05-9 38  
Cyclotrisiloxane, hexamethyl- (CAS ) \$\$ 1,1,3,3,5,5-HEXAMETHYL-CYCLOHEXASILOXANE 238257 000541-05-9 38

46 22.987 0.03 D:\MassHunter\Library\WILLEY09TH.L  
3-(2,2-dideuterobutyl)-thiophene-1,1-dioxide 111634 999111-63-4 59  
N-Methyl-2,5-dicyanopyrrole 37968 999037-96-8 53  
Pyrano[2,3-b]indole, 2,3,4,4a,9,9a-hexahydro-3-methyl- (CAS) 151866 056298-84-1 47

47 23.105 0.03 D:\MassHunter\Library\WILLEY09TH.L  
Hexadecyl hexanoate 555284 999555-28-4 25  
1,1,1,3,5,5,5-Heptamethyltrisiloxane \$\$ Bis(trimethylsiloxy)methylsiloxane 238341 001873-88-7 25

lane  
1,2-Benzisothiazol-3-amine tbdms 358158 999358-15-8 25

48 23.155 0.01 D:\MassHunter\Library\WILLEY09TH.L  
2-Methyl-5H-dibenz[b,f]azepine 199741 999199-74-1 50  
2-Naphthalene-sulfonic acid 200544 999200-54-4 50  
3,5-Dimethyl-2,6-bis(trimethylsiloxy)pyridine 411321 999411-32-1 38

49 23.283 0.87 D:\MassHunter\Library\WILLEY09TH.L  
Oleic acid, 3-hydroxypropyl ester 555014 000821-17-0 90  
\$\$ 3-Hydroxypropyl (9E)-9-octadecenoate #  
9-Octadecenoic acid (Z)-, 2-hydroxyethyl ester \$\$ Cithrol A 523343 004500-01-0 89  
9-Octadecenoic acid (Z)-, 2,3-dihydroxypropyl ester \$\$ Olein, 1-mono 587482 000111-03-5 87

50 23.430 1.02 D:\MassHunter\Library\WILLEY09TH.L  
Octadecanoic acid, 2-hydroxy-1-(hydroxymethyl)ethyl ester \$\$ Stearin, 2-mono- 591363 000621-61-4 46  
6-Octadecenoic acid, (Z)- \$\$ (6Z)- 410221 000593-39-5 44  
6-Octadecenoic acid #  
Palmitoyl chloride \$\$ Hexadecanoyl chloride- \$\$ Palmitic acid chloride 387804 000112-67-4 43

51 23.496 1.16 D:\MassHunter\Library\WILLEY09TH.L  
Tetracosanoic acid, methyl ester \$ 633616 002442-49-1 91  
\$ Methyl lignocerate \$\$ Methyl tetracosanoate  
6-Octadecenoic acid, (Z)- \$\$ (6Z)- 410221 000593-39-5 55  
6-Octadecenoic acid #  
Tetracosanoic acid, methyl ester (CAS) \$\$ Methyl lignocerate \$\$ FAME 633621 002442-49-1 53  
24:0

52 23.602 0.01 D:\MassHunter\Library\WILLEY09TH.L  
2-Methyl-5H-dibenz[b,f]azepine 199741 999199-74-1 43  
1,2-Benzisothiazol-3-amine tbdms 358158 999358-15-8 41  
1,1,1,3,5,5,5-Heptamethyltrisiloxane \$\$ Bis(trimethylsiloxy)methylsilane 238341 001873-88-7 41

53 23.972 0.02 D:\MassHunter\Library\WILLEY09TH.L  
2-Methyl-5H-dibenz[b,f]azepine 199741 999199-74-1 47  
2-Naphthalene-sulfonic acid 200544 999200-54-4 47  
1-Naphthalene-sulfonic acid 200545 999200-54-5 47

54 24.238 0.06 D:\MassHunter\Library\WILLEY09TH.L  
2-Naphthalene-sulfonic acid 200544 999200-54-4 47  
2-Methyl-5H-dibenz[b,f]azepine 199741 999199-74-1 47  
3,7,11-Trimethyltrideca-6(E),10-diene 203515 076164-11-9 35

55 24.986 0.08 D:\MassHunter\Library\WILLEY09TH.L  
Tetrasiloxane, decamethyl- (CAS) \$ 481761 000141-62-8 47



\$ Decamethyltetrasiloxane  
 Cyclotrisiloxane, hexamethyl- (CAS 238254 000541-05-9 46  
 ) \$\$ 1,1,3,3,5,5-HEXAMETHYL-CYCLOH  
 EXASILOXANE  
 Cyclotrisiloxane, hexamethyl- \$\$ D 238255 000541-05-9 46  
 imethylsiloxane cyclic trimer

56 27.032 0.05 D:\MassHunter\Library\WILLEY09TH.L  
 Ethyl 5-formyl-3-(2-ethoxycarbonyl 405830 999405-83-0 59  
 )ethyl-4-methyl-1H-pyrrole-2-carbo  
 xylate  
 2-P-NITROPHENYL-1,3,4-OXADIAZOL-5- 198194 041125-77-3 50  
 ONE \$\$ 2-p-Nitrophenyl-oxadiazol-1  
 ,3,4-one-5  
 Cyclotrisiloxane, hexamethyl- (CAS 238258 000541-05-9 46  
 ) \$\$ 1,1,3,3,5,5-HEXAMETHYL-CYCLOH  
 EXASILOXANE

57 27.408 0.02 D:\MassHunter\Library\WILLEY09TH.L  
 2-P-NITROPHENYL-1,3,4-OXADIAZOL-5- 198194 041125-77-3 50  
 ONE \$\$ 2-p-Nitrophenyl-oxadiazol-1  
 ,3,4-one-5  
 Cyclotrisiloxane, hexamethyl- (CAS 238259 000541-05-9 49  
 ) \$\$ 1,1,3,3,5,5-HEXAMETHYL-CYCLOH  
 EXASILOXANE  
 Cyclotrisiloxane, hexamethyl- (CAS 238258 000541-05-9 46  
 ) \$\$ 1,1,3,3,5,5-HEXAMETHYL-CYCLOH  
 EXASILOXANE

58 28.369 0.02 D:\MassHunter\Library\WILLEY09TH.L  
 2-P-NITROPHENYL-1,3,4-OXADIAZOL-5- 198194 041125-77-3 86  
 ONE \$\$ 2-p-Nitrophenyl-oxadiazol-1  
 ,3,4-one-5  
 Ethyl 5-formyl-3-(2-ethoxycarbonyl 405830 999405-83-0 59  
 )ethyl-4-methyl-1H-pyrrole-2-carbo  
 xylate  
 Cyclotrisiloxane, hexamethyl- (CAS 238259 000541-05-9 49  
 ) \$\$ 1,1,3,3,5,5-HEXAMETHYL-CYCLOH  
 EXASILOXANE

59 29.669 0.05 D:\MassHunter\Library\WILLEY09TH.L  
 Ethyl 5-formyl-3-(2-ethoxycarbonyl 405830 999405-83-0 59  
 )ethyl-4-methyl-1H-pyrrole-2-carbo  
 xylate  
 2-P-NITROPHENYL-1,3,4-OXADIAZOL-5- 198194 041125-77-3 50  
 ONE \$\$ 2-p-Nitrophenyl-oxadiazol-1  
 ,3,4-one-5  
 Tetrasiloxane, decamethyl- (CAS) \$ 481761 000141-62-8 47  
 \$ Decamethyltetrasiloxane

UMUM.M Tue Nov 01 15:53:21 2022

**INSTRUMENT CONTROL PARAMETERS: GCMS**

D:\MassHunter\GCMS\1\methods\UMUM.M  
 Tue Nov 01 15:45:43 2022

Control Information

-----  
Sample Inlet : GC  
Injection Source : GC ALS  
Injection Location: Front  
Mass Spectrometer : Enabled

## GC

### GC Summary

Run Time 39 min  
Post Run Time 0 min

## OVEN

### Temperature

Setpoint On  
(Initial) 50 °C  
Hold Time 0 min  
Post Run 70 °C  
Program  
#1 Rate 10 °C/min  
#1 Value 290 °C  
#1 Hold Time 15 min

Equilibration Time 0.25 min  
Max Temperature 325 °C  
Maximum Temperature Override Disabled  
Slow Fan Disabled

## FRONT SS INLET HE

Mode Split  
Heater On 290 °C  
Pressure On 7.6522 psi  
Total Flow On 104 mL/min  
Septum Purge Flow On 3 mL/min  
Gas Saver On 20 After 3 min mL/min  
Split Ratio 100 :1  
Split Flow 100 mL/min  
Liner A Liner has not been selected.  
Vol Injek 0,2 µL

## THERMAL AUX 2 (MSD TRANSFER LINE)

Temperature  
Setpoint On  
(Initial) 290 °C

## COLUMN

Column #1  
Flow  
Setpoint Off  
(Initial) 1 mL/min  
Post Run 0.57353 mL/min

Column Information Agilent 19091S-433UI  
HP-5ms Ultra Inert  
Temperature Range -60 °C—325 °C (350 °C)  
Dimensions 30 m x 250 µm x 0.25 µm  
Column lock Unlocked  
In Front SS Inlet He

Out MSD  
(Initial) 50 °C  
Pressure 7.6522 psi  
Flow 1 mL/min  
Average Velocity 36.445 cm/sec  
Holdup Time 1.3719 min  
Control Mode Constant Flow

Column Outlet Pressure 0 psi

Valve 1  
Name ?  
Type Gas Sampling Valve  
GSV Loop Volume 1 mL  
Load Time 0.5 min  
Inject Time 0.5 min

Signals  
Signal #1: Test Plot  
Description Test Plot  
Save Off  
Data Rate 50 Hz  
Dual Injection Assignment Front Sample

Signal #2:  
Description None

Signal #3:  
Description None

Signal #4:  
Description None

## **MS INFORMATION**

### General Information

-----

Acquisition Mode : Scan  
Solvent Delay (minutes) : 2  
Tune file : D:\MassHunter\GCMS\1\5977\BBM etune.u  
EM Setting mode Delta : 0.000000

Normal or Fast Scanning : Normal Scanning  
Trace Ion Detection : Off  
Run Time (if MS only) : 26.67 minutes

[Scan Parameters]  
Start Time : 2  
Low Mass : 35  
High Mass : 650  
Threshold : 150  
A/D Samples: : 4

[MSZones]  
MS Source : 230 C maximum 250 C  
MS Quad : 150 C maximum 200 C

Timed Events

-----  
Number Events= 0

**TUNE PARAMETERS for SN: US1904M017**  
-----

Trace Ion Detection is OFF.

34.593 : EMISSION  
70.007 : ENERGY  
0.503 : REPELLER  
89.822 : IONFOCUS  
17.627 : ENTRANCE\_LENS  
1408.533 : EMVOLTS  
1408.5 : Actual EMV  
0.33 : GAIN FACTOR  
2171.000 : AMUGAIN  
139.750 : AMUOFFSET  
2.000 : FILAMENT  
0.000 : DCPOLARITY  
10.278 : ENTLENSOFFSET  
8.002 : Ion\_Body  
-0.398 : EXTLENS  
-5.000 : MASSGAIN  
-23.000 : MASSOFFSET

END OF INSTRUMENT CONTROL PARAMETERS  
-----

- **XRD**

- **Sebelum Kalsinasi**

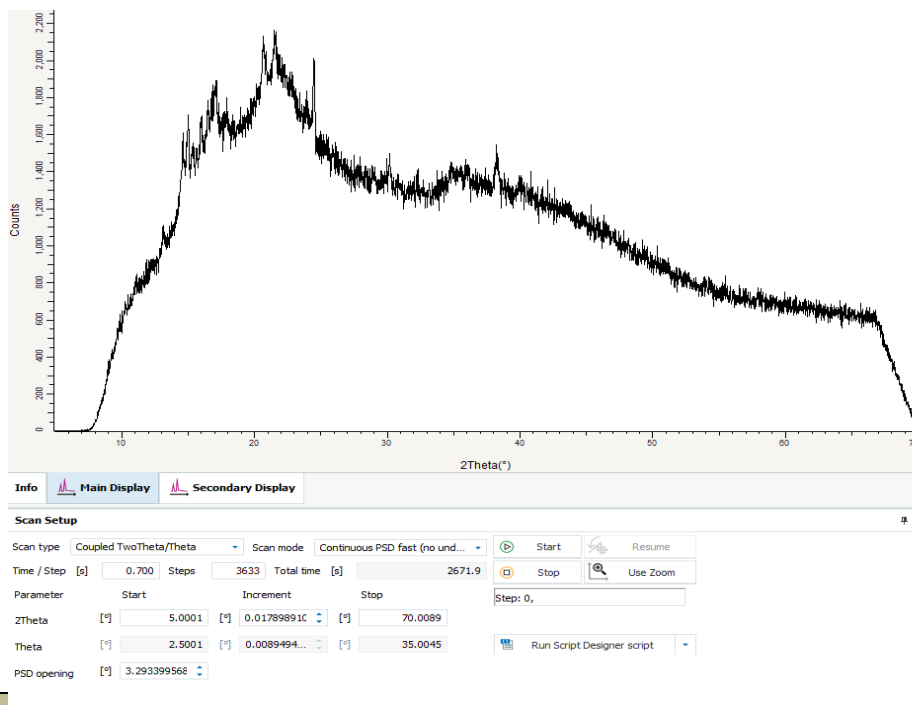
**Test Report**

<b>Name of Project</b>	XRD 054 – Kulit Pisang Sebelum Reaksi/ Sebelum Kalsinasi
<b>User Name</b>	M. Alfian Pratama
<b>Name of the Test</b>	XRD
<b>Test Date</b>	31 Oktober 2022

**Test Runs Chart**



## Test Runs Chart



## Validation

Chair of Integrated Laboratory



Dr. Didied Haryono, ST., MT.

NIP. 196810062001121002

Head of Nanoparticles Laboratory



Dr. Lusiani Dewi Assaat, S.Pd., M.Si.

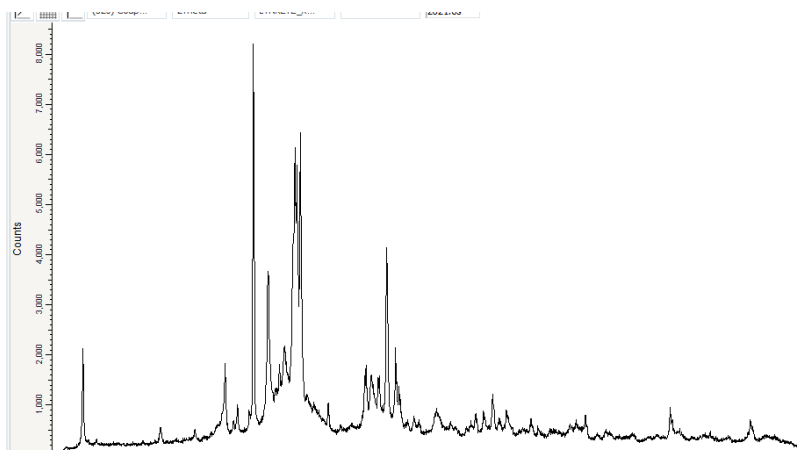
NIP. 198010102005012003

- **Setelah Kalsinasi**

## Test Report

<b>Name of Project</b>	XRD 40 – 550 5 Jam
<b>User Name</b>	M. Alfian Pratama
<b>Name of the Test</b>	XRD
<b>Test Date</b>	12 Oktober 2022

### Test Runs Chart



## Validation

Chair of Integrated Laboratory



Dr. Didied Haryono, ST., MT.

NIP. 196810062001121002

Head of Nanoparticles Laboratory



• S

Dr. Lusiani Dewi Assaat, S.Pd., M.Si.

NIP. 198010102005012003

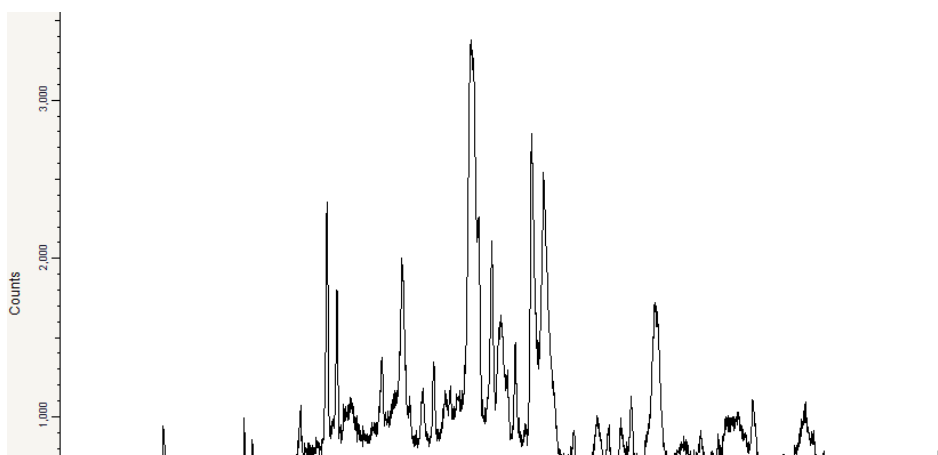
etelah Reaksi  
Transesterifikas

i

## Test Report

<b>Name of Project</b>	XRD 055 – Kulit Pisang Sesudah Reaksi
<b>User Name</b>	M. Alfian Pratama
<b>Name of the Test</b>	XRD
<b>Test Date</b>	31 Oktober 2022

## Test Runs Chart



## Validation

Chair of Integrated Laboratory



Dr. Didied Haryono, ST., MT.

NIP. 196810062001121002

Head of Nanoparticles Laboratory



Dr. Lusiani Dewi Assaat, S.Pd., M.Si.

NIP. 198010102005012003

- SEM

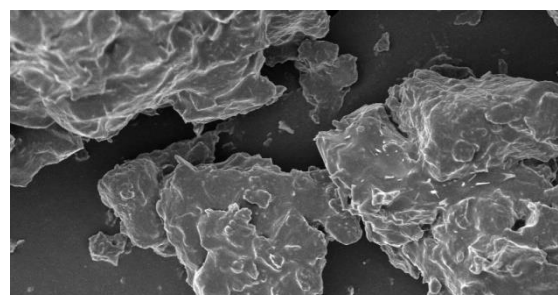
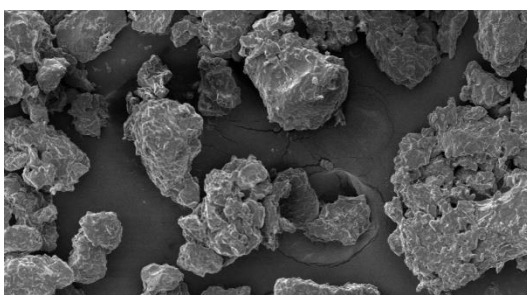
- Sebelum Kalsinasi dan Setelah Kalsinasi

## ● Test Report

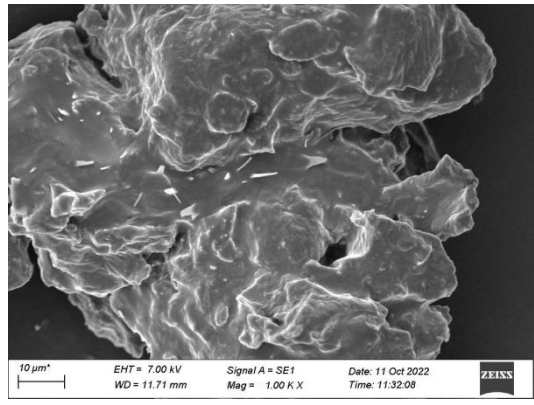
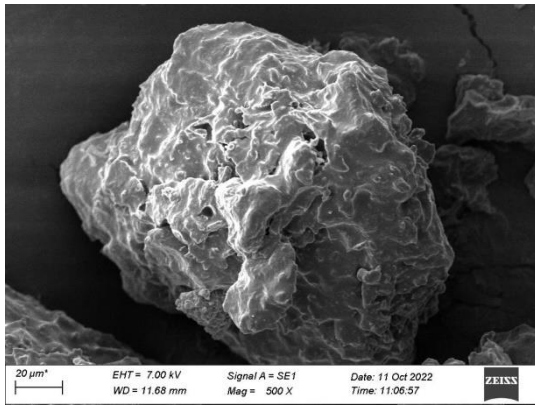
Name of Project	SEM (052 – 053) – Kitosan Kulit Pisang
User Name	Safitri Mukti Rahayu
Name of the Test	SEM
Test Date	11 Oktober 2022

## Test Runs Image

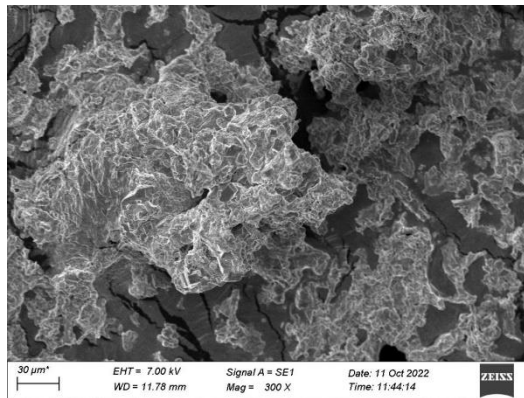
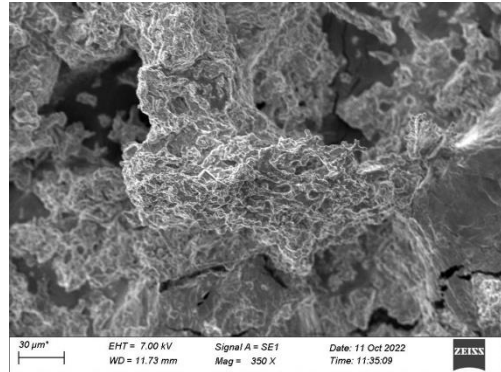
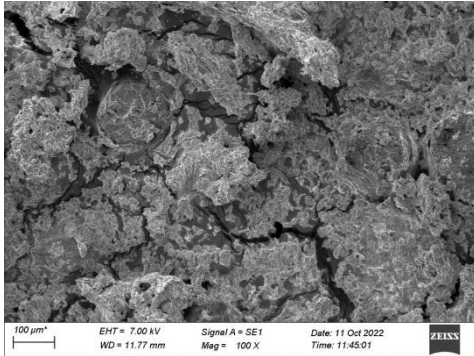
- SEBELUM KALSINASI







● SETELAH KALSINASI



Validation

Chair of Integrated Laboratory

Dr. Didied Haryono, ST., MT.

NIP. 196810062001121002

Head of Nanomaterials Laboratory

Dr. Lusiani Dewi Assaat, S.Pd., M.Si.

NIP. 198010102005012003

- EDX

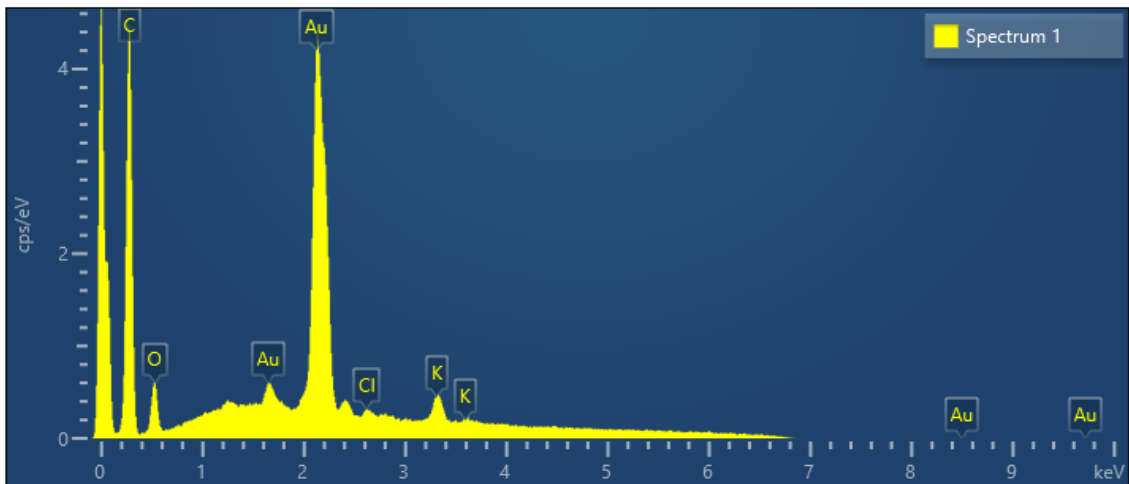
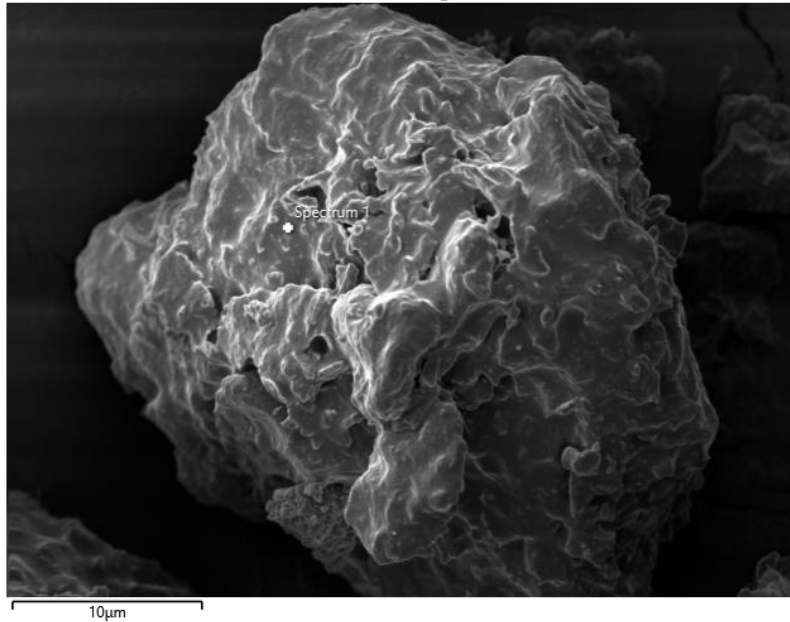
- Sebelum Kalsinasi

### Test Report

Name of Project	SEM 052 – Chi Sebelum Kalsinasi
User Name	Safitri Mukti Rahayu
Name of the Test	EDS
Test Date	11 Oktober 2022

#### Test Runs Image

Electron Image 1



Spectrum 1				
------------	--	--	--	--

Element	Line Type	Weight %	Weight % Sigma	Atomic %
C	K series	77.76	0.51	87.84
O	K series	8.74	0.24	7.41
Cl	K series	1.85	0.25	0.71
K	K series	11.64	0.45	4.04
Total		100.00		100.00

**Validation**

Chair of Integrated Laboratory



Dr. Didied Haryono, ST., MT.

NIP. 196810062001121002

Head of Nanoparticles Laboratory



Dr. Lusiani Dewi Assaat, S.Pd., M.Si.

NIP. 198010102005012003

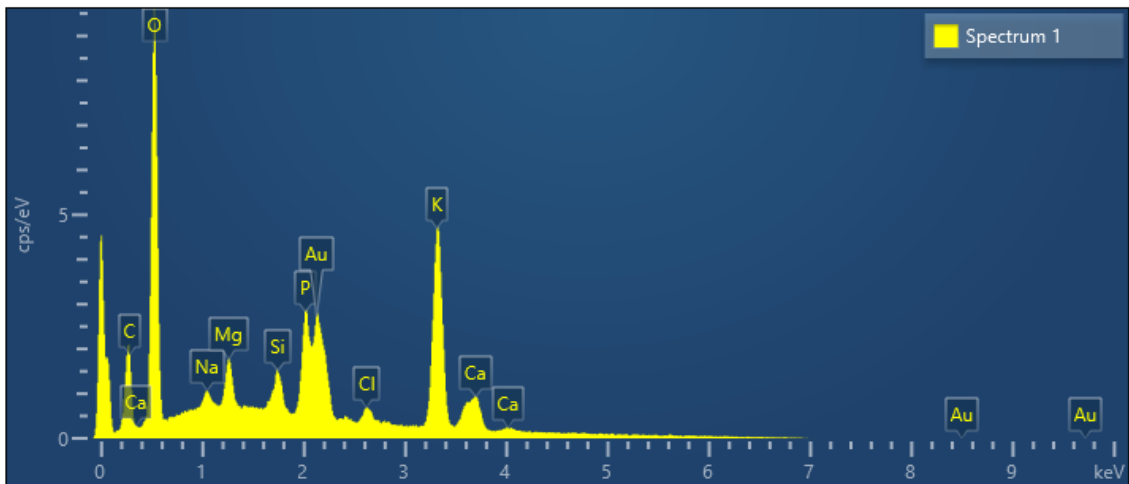
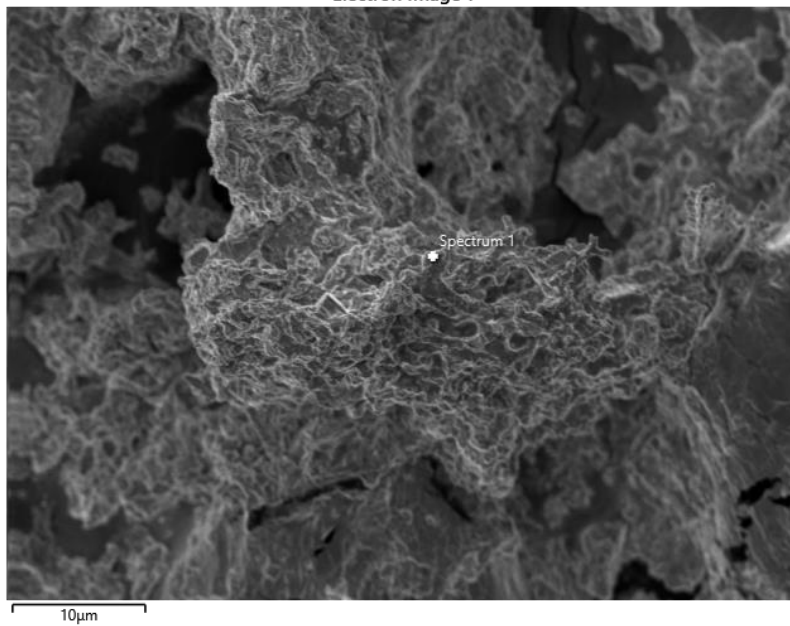
- **Setelah Kalsinasi**

## Test Report

<b>Name of Project</b>	SEM 053 – Chi Setelah Kalsinasi
<b>User Name</b>	Safitri Mukti Rahayu
<b>Name of the Test</b>	EDS
<b>Test Date</b>	11 Oktober 2022

### Test Runs Image

Electron Image 1



Spectrum 1				
Element	Line Type	Weight %	Weight % Sigma	Atomic %
C	K series	6.57	0.34	12.96

O	K series	32.22	0.26	47.74
Na	K series	0.53	0.06	0.55
Mg	K series	1.67	0.06	1.63
Si	K series	1.79	0.07	1.51
P	K series	6.09	0.13	4.66
Cl	K series	1.55	0.10	1.04
K	K series	40.09	0.31	24.30
Ca	K series	9.50	0.27	5.62
Total		100.00		100.00

### Validation

Chair of Integrated Laboratory



Dr. Didied Haryono, ST., MT.

NIP. 196810062001121002

Head of Nanoparticles Laboratory



Dr. Lusiani Dewi Assaat, S.Pd., M.Si.

NIP. 198010102005012003

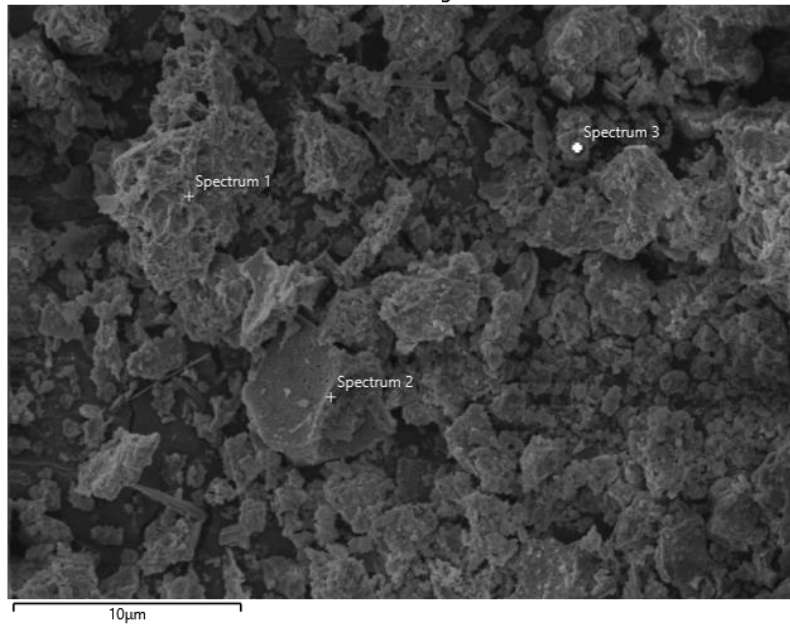
- **Setelah Reaksi Transesterifikasi**

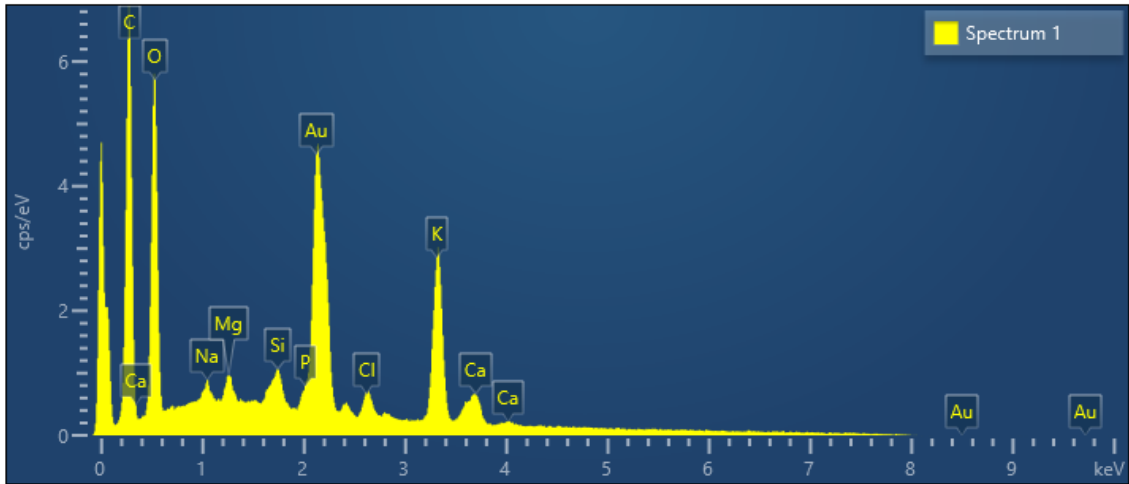
## Test Report

<b>Name of Project</b>	SEM 058 – Kitosan Kulit Pisang Setelah Transesterifikasi
<b>User Name</b>	Safitri Mukti Rahayu
<b>Name of the Test</b>	EDS
<b>Test Date</b>	25 Oktober 2022

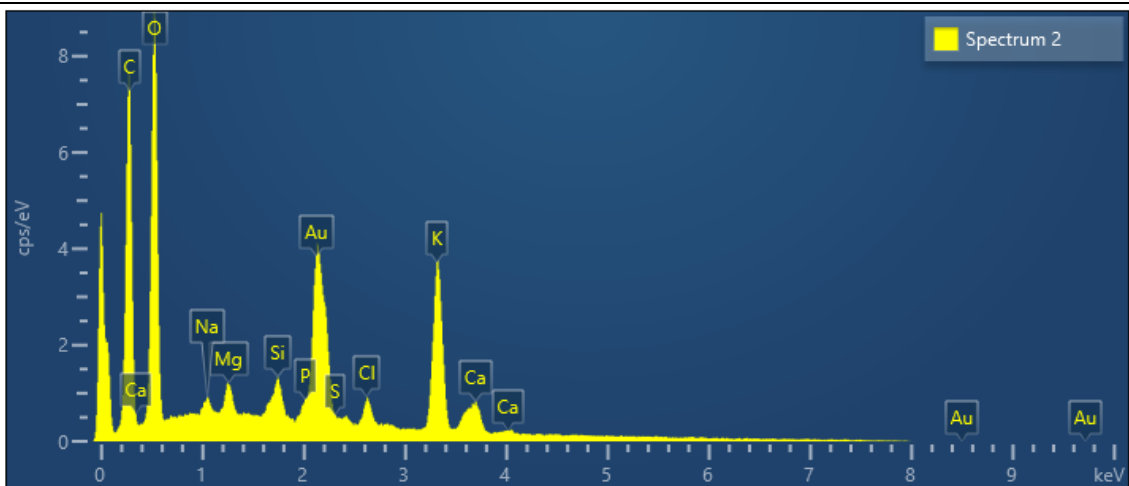
### Test Runs Image

Electron Image 1





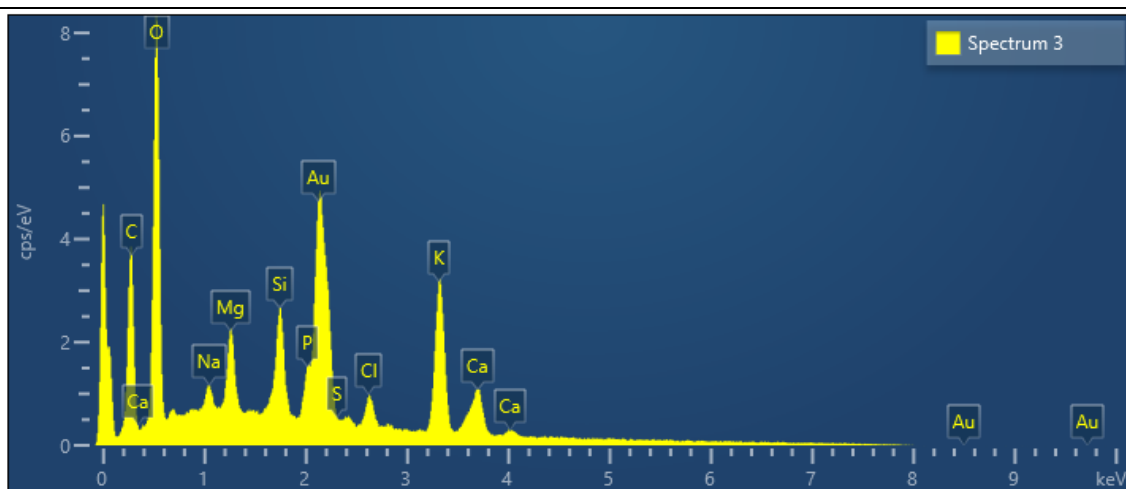
Spectrum 1				
Element	Line Type	Weight %	Weight % Sigma	Atomic %
C	K series	40.36	0.43	56.90
O	K series	26.49	0.28	28.03
Na	K series	0.54	0.06	0.40
Mg	K series	0.75	0.05	0.52
Si	K series	1.25	0.07	0.75
P	K series	0.92	0.11	0.51
Cl	K series	1.97	0.10	0.94
K	K series	22.36	0.26	9.68
Ca	K series	5.36	0.20	2.26
Total		100.00		100.00



Spectrum 2				
Element	Line Type	Weight %	Weight % Sigma	Atomic %
C	K series	35.18	0.43	50.67
O	K series	31.05	0.29	33.57
Na	K series	0.50	0.05	0.38
Mg	K series	0.92	0.05	0.66
Si	K series	1.35	0.06	0.83
P	K series	0.86	0.10	0.48
S	K series	0.37	0.11	0.20
Cl	K series	2.03	0.09	0.99
K	K series	22.46	0.24	9.94



Ca	K series	5.27	0.18	2.28
Total		100.00		100.00



Spectrum 3				
Element	Line Type	Weight %	Weight % Sigma	Atomic %
C	K series	23.46	0.54	37.29
O	K series	32.77	0.34	39.11
Na	K series	0.84	0.07	0.70
Mg	K series	2.53	0.07	1.98
Si	K series	4.18	0.09	2.84
P	K series	2.54	0.14	1.56
S	K series	0.44	0.14	0.26
Cl	K series	2.59	0.11	1.39
K	K series	21.85	0.27	10.67
Ca	K series	8.79	0.23	4.19
Total		100.00		100.00

**Validation**

Chair of Integrated Laboratory



Dr. Didied Haryono, ST., MT.

NIP. 196810062001121002

Head of Nanoparticles Laboratory

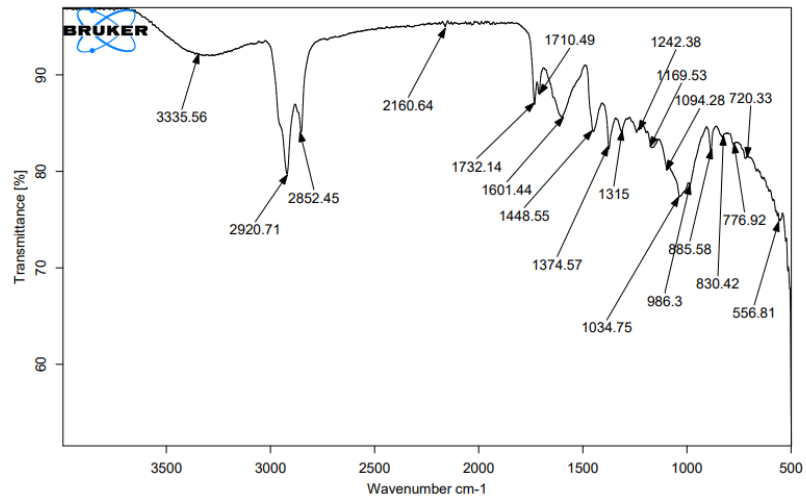
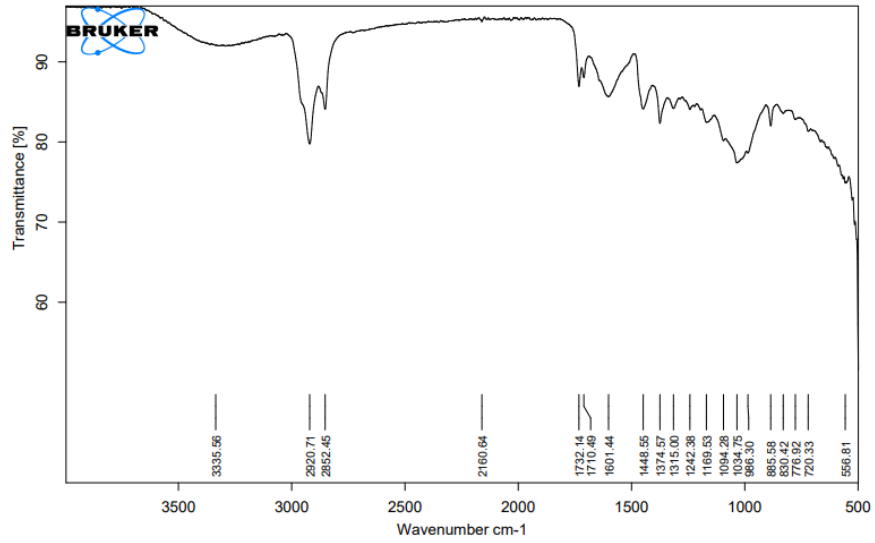


Dr. Lusiani Dewi Assaat, S.Pd., M.Si.

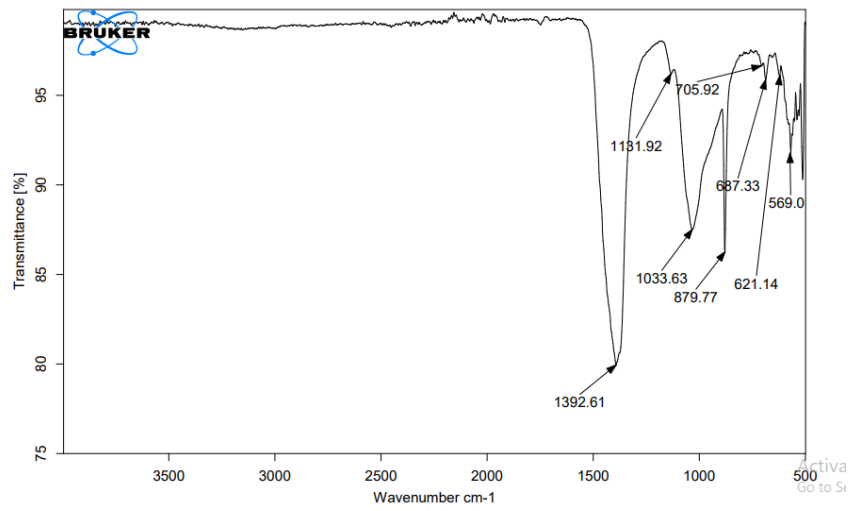
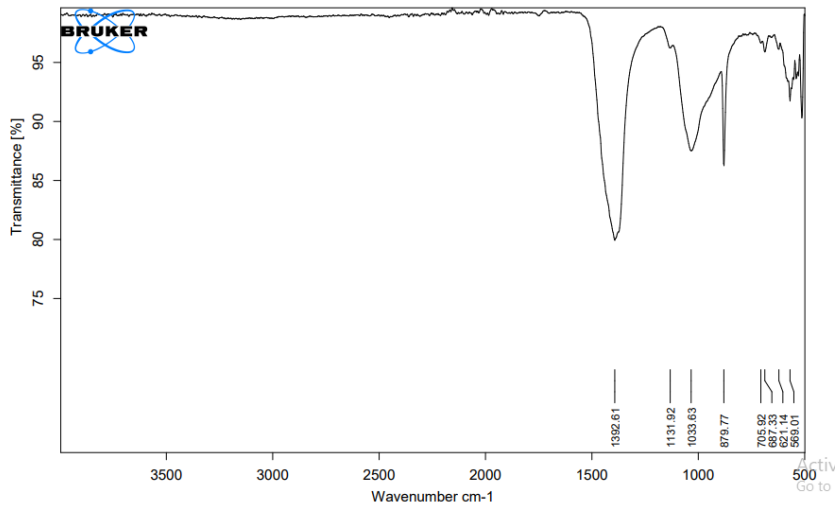
NIP. 198010102005012003

- FTIR

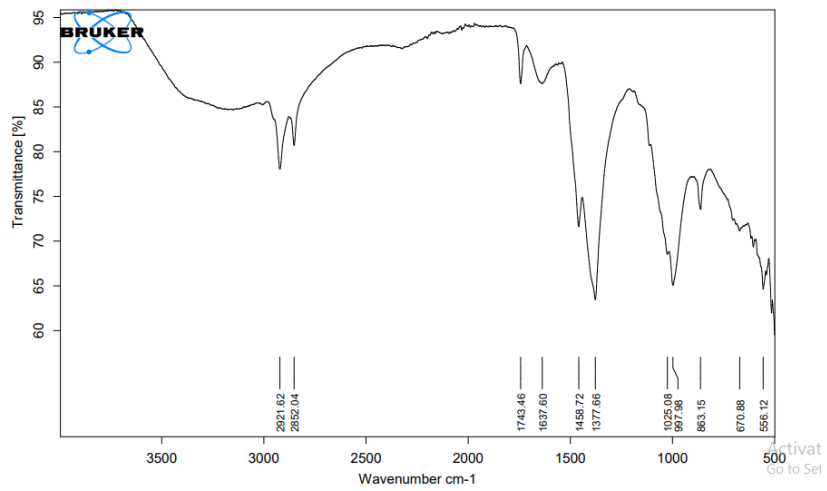
• Sebelum Kalsinasi

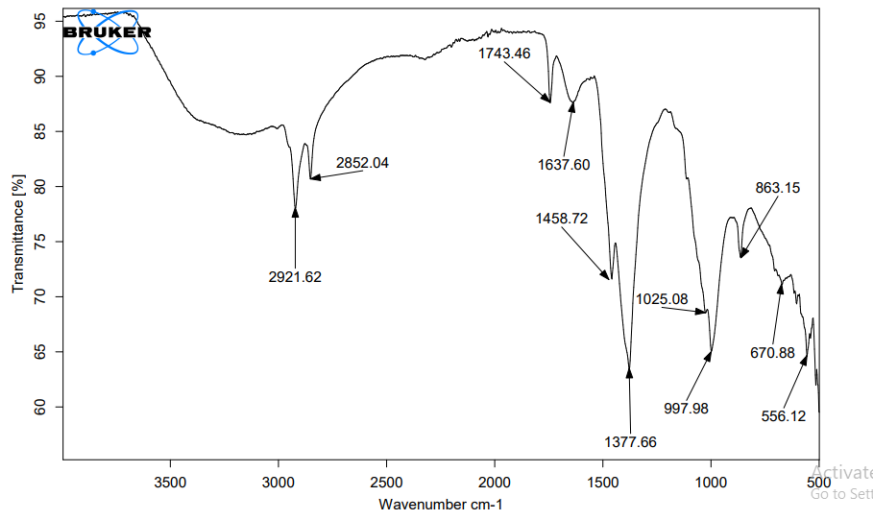


- **Setelah Kalsinasi**



- **Setelah Reaksi Transesterifikasi**





### C. DOKUMENTASI

#### ❖ Proses Pembuatan Katalis



(a)



(b)



(c)



(d)

**Gambar 1.** Preparasi Katalis Kulit Pisang Kepok Matang (a) Menimbang kulit pisang yang dibutuhkan (b) Pencucian (c) Pemetongan (d) Pengeringan dengan oven



(a)



(b)



(c)



(d)



(e)



(f)

**Gambar 2.** Preparasi Kalsinasi Kulit Pisang Kepok Matang (a) Penimbangan hasil oven (b) Pengayakan (c) Pembagian pada cawan porselen (d) Katalis 450°C (e) Katalis 550°C (f) Katalis 650°C

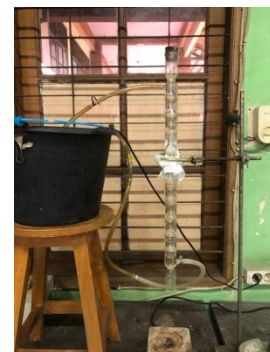
### ❖ Proses Transesterifikasi



(a)



(b)



(c)



(d)



(e)



(f)



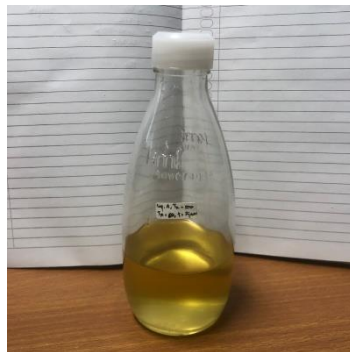
(g)



(h)



(i)



(j)

**Gambar 3.** Proses Transesterifikasi (a) Pengukuran minyak dan methanol (b) Es batu (c) Rangkaian alat (d) Methanol dan katalis di panaskan (e) Minyak dimasukkan (f) Rangkaian sempurna (g,h,i) Penyaringan Minyak (j) Hasil biodiesel





(a)



(b)



(c)

**Gambar 4.** Reusability Katalis (a) Katalis sehabis pemakaian (b) Proses pemanasan (c) Penyaringan

❖ **Analisa Kimia & Fisik**



(a)



(b)

**Gambar 5.** Uji FFA (a) Sebelum titrasi (b) Setelah titrasi

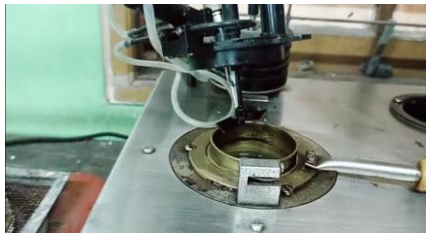


(a)



(b)

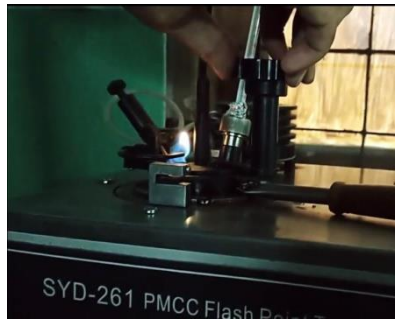




(c)



(d)

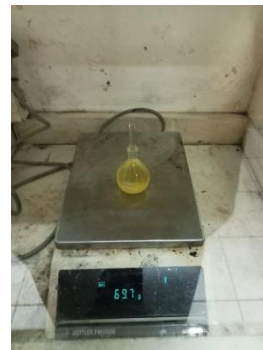


(e)

**Gambar 6.** Analisa Flash Point (a) ON alat (b) Pemasangan gas (c) Pemasukan sampel (d) Menyalakan pemantik (e) Keluar titik nyala



(a)



(b)

**Gambar 7.** Analisa (a) Viskositas (b) Densitas

**KEMENTERIAN PENDIDIKAN DAN KEBUDAYAAN  
JURUSAN TEKNIK KIMIA- FAKULTAS TEKNIK UNIVERSITAS  
SULTAN AGENG TIRTAYASA**

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**BORANG REVISI LAPORAN PENELITIAN**

Judul Penelitian : Potensi Kulit Pisang Kepok Matang Sebagai Katalis Heterogen  
Penghasil Biodiesel dengan Bahan Baku Minyak Jelantah

Nama Mahasiswa : 1. Nisrina Nada Salma 3335180038

2. Umi Hanifah 3335180028

**REVISI**

1. Peraturan tentang biodiesel seperti sudah ada yang baru, mohon direvisi yang peraturan ESDM No 12 tahun 2015.
2. Penelitian tentang kuliah pisang kepok hasilnya belum disampaikan pada halaman 2 di bagian latar belakang. Ini bisa ditambahkan bagian dari state of the art dari penelitian ini termasuk beberapa penelitian pengambilan kalium dari bahan yang lain dan bagaimana hasil kualitas biodiesel yang dihasilkan
3. Karakteristik dari kalium untuk berbagai variasi sebaiknya dilengkapi pada bagian awal sehingga nanti terpilih yang terbaik
4. Analisa pengaruh dari beberapa parameter perlu diperkuat sehingga terlihat mana faktor yang paling dominan berpengaruh terhadap yield

Cilegon, 4 Januari 2023

Dosen Penguji 1,



Dr. Ing-Anton Irawan, ST., MT  
NIP.197510012008011007

**KEMENTERIAN PENDIDIKAN DAN KEBUDAYAAN  
JURUSAN TEKNIK KIMIA- FAKULTAS TEKNIK  
UNIVERSITAS SULTAN AGENG TIRTAYASA**

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**BORANG REVISI LAPORAN PENELITIAN**

Judul Penelitian : Potensi Kulit Pisang Kepok Matang Sebagai Katalis Heterogen Penghasil Biodiesel dengan Bahan Baku Minyak Jelantah

Nama Mahasiswa : 1. Nisrina Nada Salma                   3335180038  
                      2. Umi Hanifah                                   3335180028

<b>REVISI</b>

Cilegon, 10 Januari 2022  
Dosen Penguji 2,

Wardalia, ST., MT  
NIP.198406202008122002